

# Year 13 Chemistry: Academic Calendar 20-21

## Term 1

	Week 1 (2 days)	Week 2	Week 3	Week 4	Week 5	Week 6	Week 7	Week 8 (3 days)
<b>Teacher 1</b>		5.1.1 Orders, rate equations and rate constants	5.1.1 Concentration-time graphs. Rate-concentration graphs.	5.1.1 Rate determining steps. Rate constants and T	5.1.2 Kc recap. Kc part 2	5.1.2 Kp. Controlling the position of equilibrium	5.1.3 Bronsted-Lowry acids and bases. pH scale and strong acids	5.1.3 Ka. pH of weak acids
<b>Teacher 2</b>		Organic mechanism review	6.1.1 Benzene intro Electrophilic substitution with benzene	6.1.1 Phenol. Distribution and directing groups	6.1.2 Carbonyl compounds. Identifying aldehydes and ketones	6.1.3 Carboxylic acids.	6.1.3 Carboxylic acids and derivatives	6.2.1 Amines
<b>Practical</b>			PAG 10		PAG 7	PAG 8		
<b>Assessments</b>			AS content assessment 21/09		Chapter 18 and 19 and module 1-3		Chapter 25 and 26 and module 1,2,4	
<b>Homework</b>	Chapter 18 & 16	Chapter 18 & 25	Chapter 18 & 26	Chapter 19 & 26	Chapter 19 & 26	Chapter 20 & 26	Chapter 20 & 27	Chapter 20 & 27

## Term 2

	Week 1 (4 days)	Week 2	Week 3	Week 4	Week 5	Week 6	Week 7
<b>Teacher 1</b>	5.1.3 pH strong bases	5.1.3 Buffer solutions	5.1.3 Buffer solutions in the body Neutralisation	5.2.1 Lattice enthalpy Enthalpy changes in solution	5.2.1 Factors affecting lattice enthalpy and hydration		
<b>Teacher 2</b>	6.2.2 Amino acids, amides and chirality	6.2.3 Condensation polymers	Unified chemistry synoptic review	Unified chemistry synoptic review	Unified chemistry synoptic review		
<b>Practical</b>	PAG 11						
<b>Assessments</b>						Year 13 Mock Exams	
<b>Homework</b>	Chapter 21 & 27	Revisited homework	Chapter 22 & 28	Chapter 22	Chapter 22		

## Term 3

	Week 1 (4 days)	Week 2	Week 3	Week 4	Week 5	Week 6
<b>Teacher 1</b>	5.2.2 Entropy Free Energy	5.2.3 Redox reactions. Manganate(VII) titration	5.2.3 Iodine/thiosulphate redox titration.	5.2.3 Electrode potentials. Predictions from electrode potentials	5.2.3 Storage and fuel cells. 5.3.1 d-block elements	5.3.1 Formations and shapes of complex ions Stereoisomerism in complex ions
<b>Teacher 2</b>	6.2.4 Carbon-carbon bond formation	6.2.5 Further synthesis routes	6.3.1 Chromatography and functional group analysis	6.3.2 NMR spectroscopy Carbon 13 NMR spectroscopy	6.3.2 Proton NMR Spectroscopy Interpreting NMR spectroscopy	Unified chemistry synoptic review
<b>Practical</b>					PAG 12	PAG catch up
<b>Assessments</b>	Chapter 21 and 22 and module 1-3				Chapter 28 and 29 and module 1,2,4	
<b>Homework</b>	Revisited homework	Chapter 23 & 29	Chapter 23 & 29	Chapter 23 & 29	Chapter 23 & 29	Chapter 24

## Term 4

	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6 (4 days)
<b>Teacher 1</b>	5.3.1 Ligand substitution and	5.3.1 Redox and qualitative analysis	Unified chemistry synoptic review	Exam technique Revision	Exam technique Revision	Exam technique Revision
<b>Teacher 2</b>	6.3.2 Combined spectroscopy	6.3.2 Combined spectroscopy	Unified chemistry synoptic review	Exam technique Revision	Exam technique Revision	Exam technique Revision
<b>Assessments</b>		Mock				
<b>Homework</b>	Revisited homework	Chapter 24				

## Term 5

	Week 1	Week 2	Week 3 (4 days)	Week 4	Week 5	Week 6
<b>Teacher 1</b>			Exam technique Revision	Exam technique Revision	Exam technique Revision	Exam technique Revision
<b>Teacher 2</b>			Exam technique Revision	Exam technique Revision	Exam technique Revision	Exam technique Revision
<b>Assessments</b>	Y13 Mocks					
<b>Homework</b>						

## Term 6

	Week 1	Week 2	Week 3	Week 4	Week 5	Week 6	Week 7 3 teaching days
<b>Teacher 1</b>	Paper 1 1h30 Periodic table, Elements and physical Chemistry (Module 1,2,3,5) 7 <sup>th</sup> June 2021						
<b>Teacher 2</b>	Paper 2 1h30 Synthesis and Analytical Techniques (Module 1,2,4,6) 11 <sup>th</sup> June 2021						
<b>Assessments</b>							
<b>Homework</b>	Paper 3 2h15 Unified Chemistry (Module 1,2,3,4,5,6) 18 <sup>th</sup> June 2021						